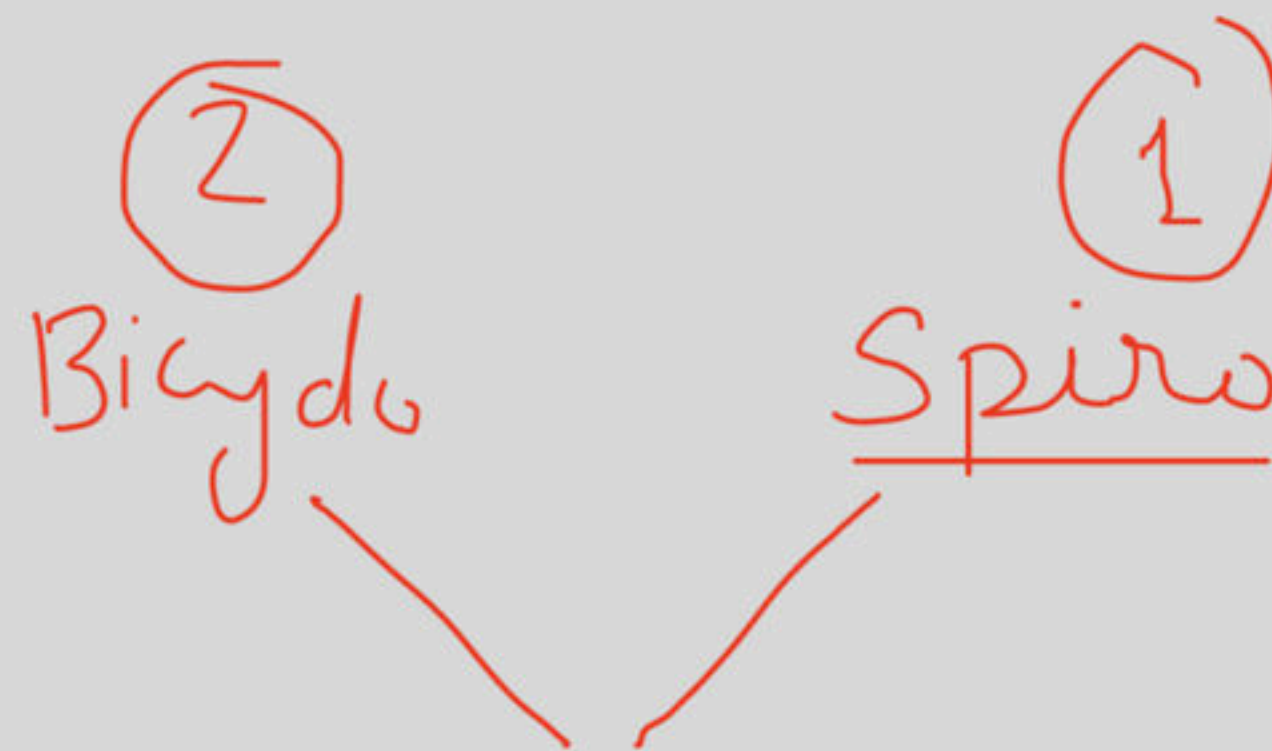


# IUPAC Nomenclature of Bicyclic System

Special class



# **INTRODUCTION TO GENERAL ORGANIC CHEMISTRY**



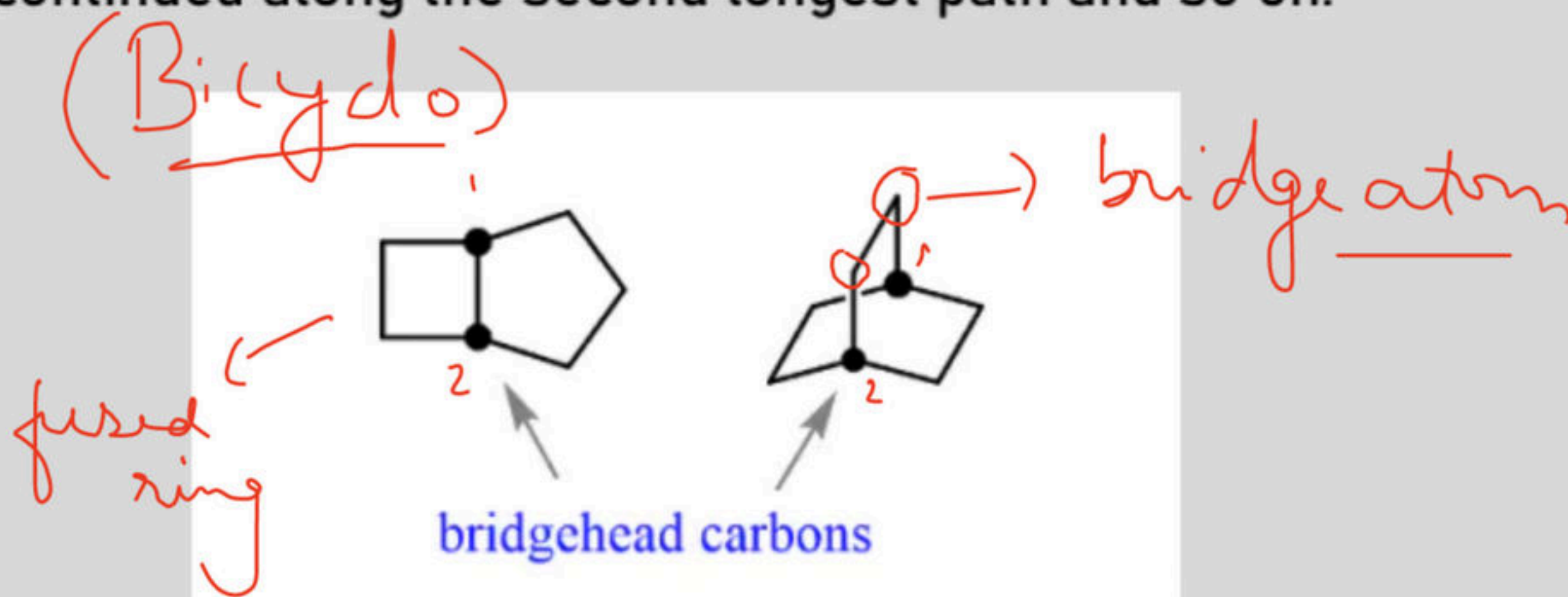
## IUPAC Nomenclature of Bicyclic System



2 ring system  
—

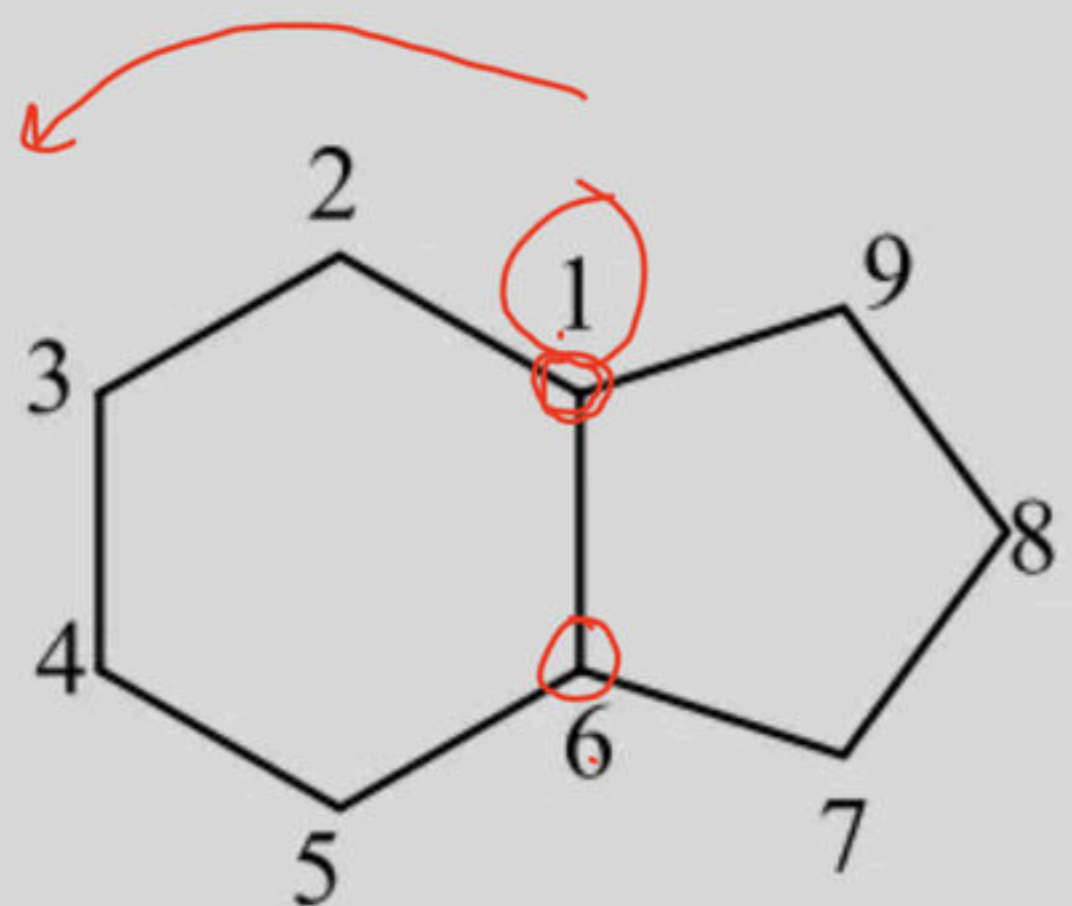
✓ (GPAT 2014)

1. The total number of atoms in all rings together gives the root name of the compound and sometimes. It is followed by a suffix denoting the functional group with the highest priority.
2. The carbon chain numbering always begins at one bridgehead atom (i.e., where the rings meet), and then it continues along the carbon chain along the longest path, to the next bridgehead atom. Numbering is then continued along the second longest path and so on.

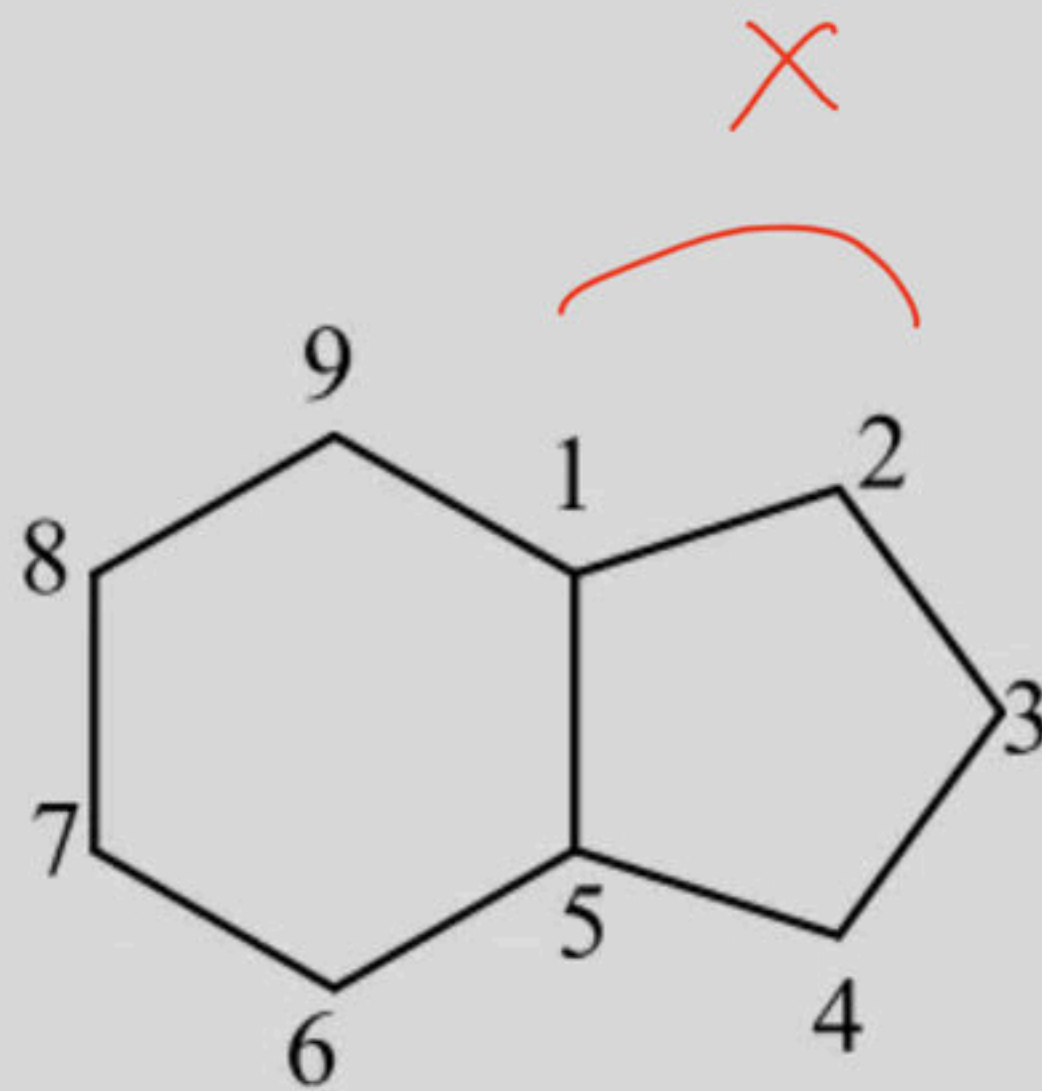




Bicyclo

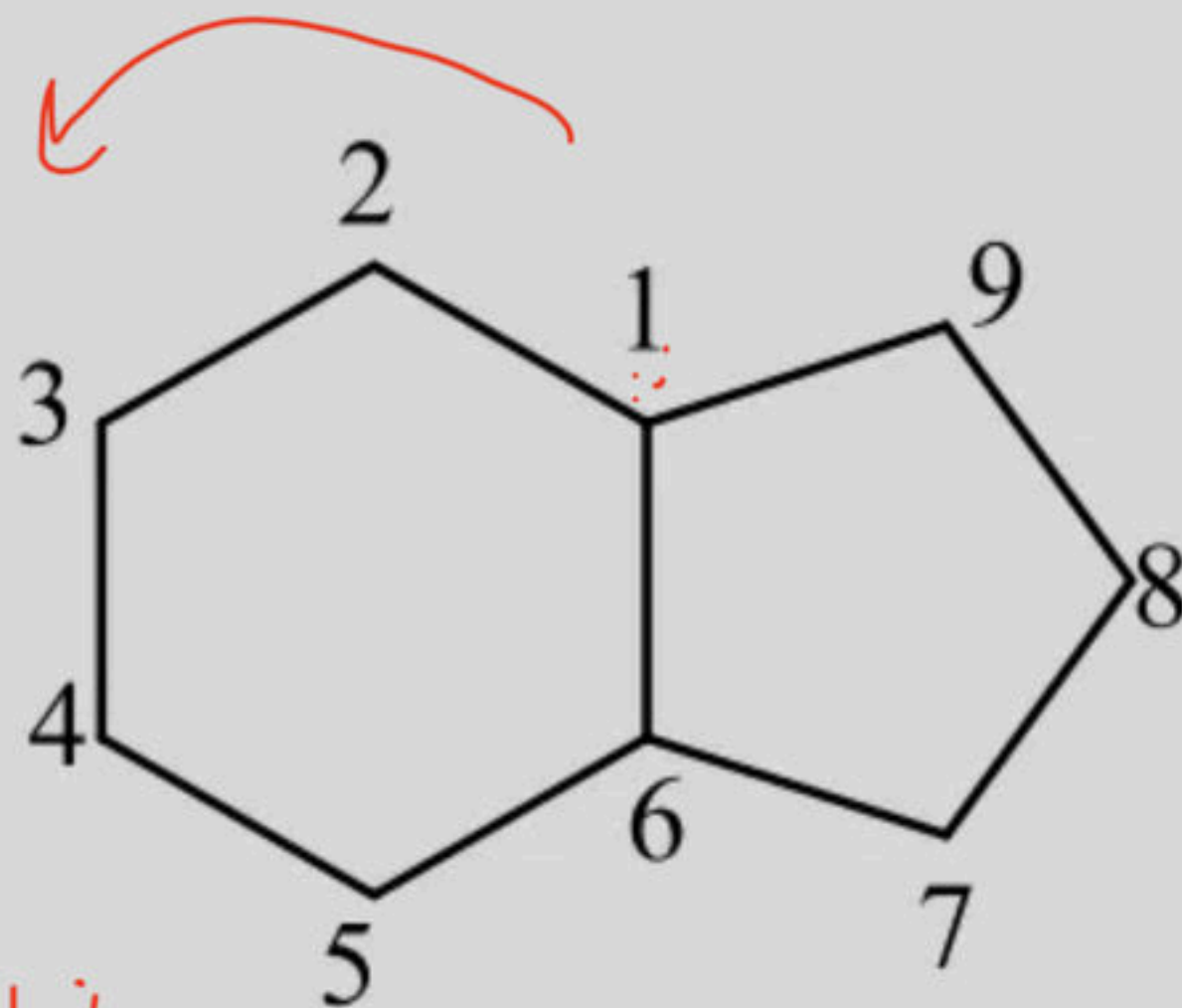


OR



start from fused carbon  $\rightarrow$  toward bigger ring.

## Correct Numbering



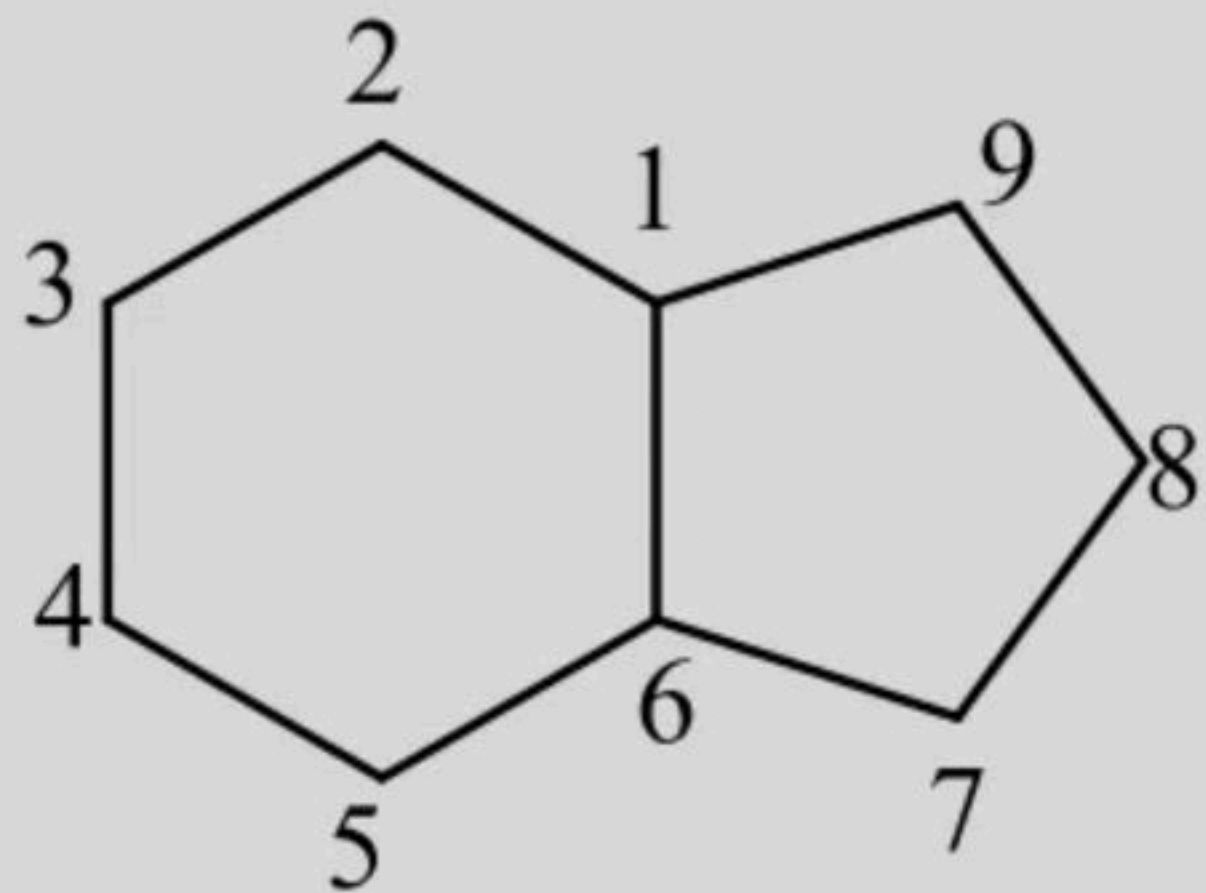
✓ Substituent

✓ Bicyclo  
[ ]

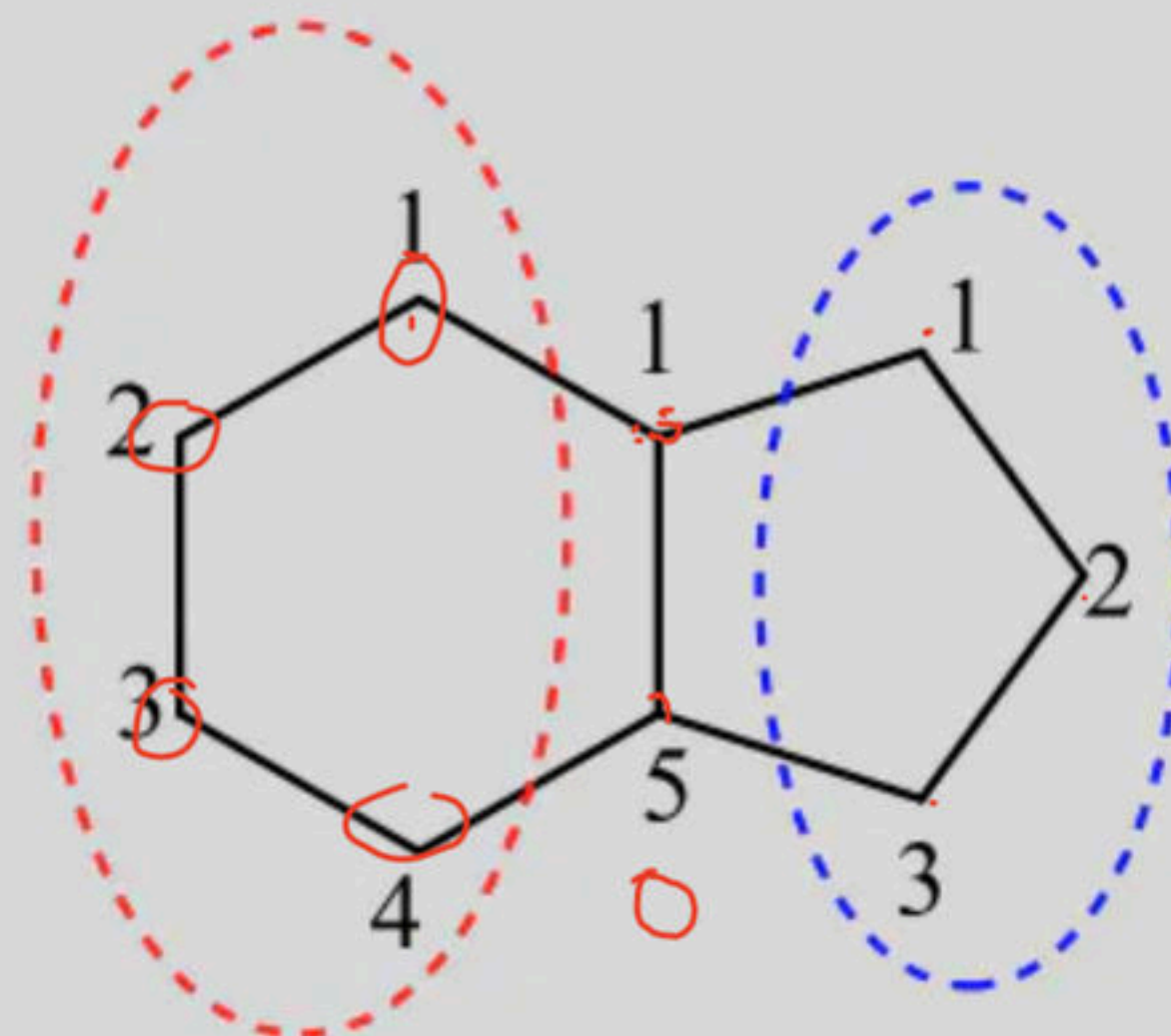
• Total no. of C

alkane

↓  
Total no. of carbon  
atom



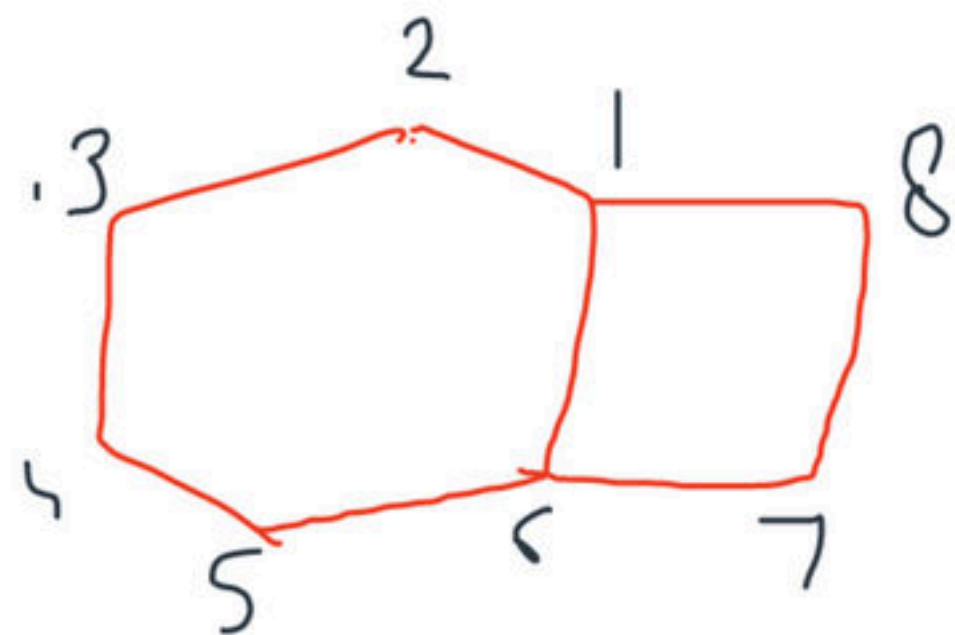
✓ nonane



4 carbon ring

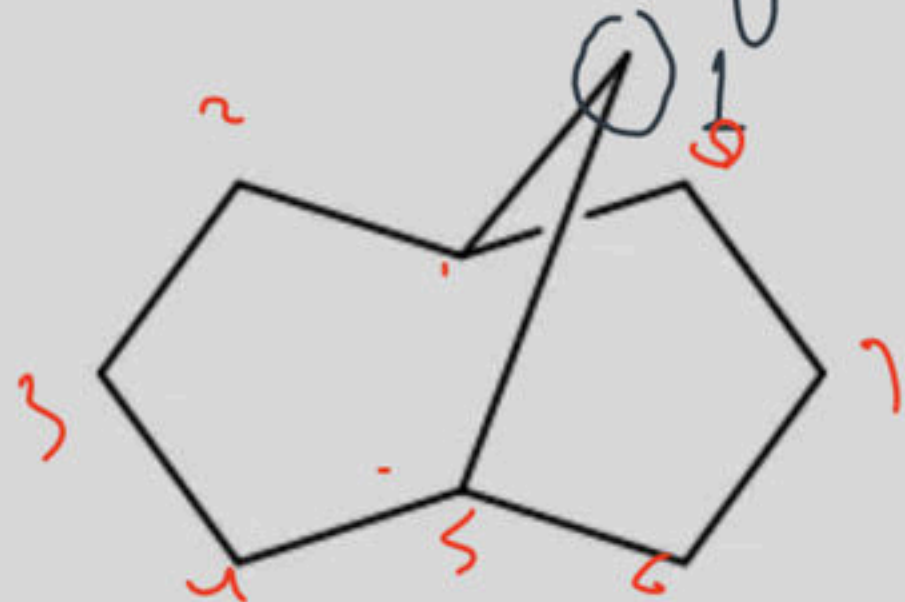
3 carbon ring

Bicyclo[4.3.0]nonane

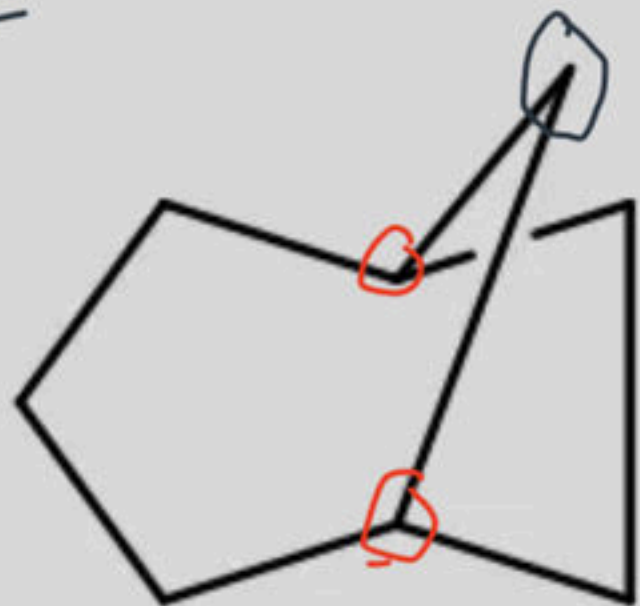


✓ Bicyclo [4.2.0] octane

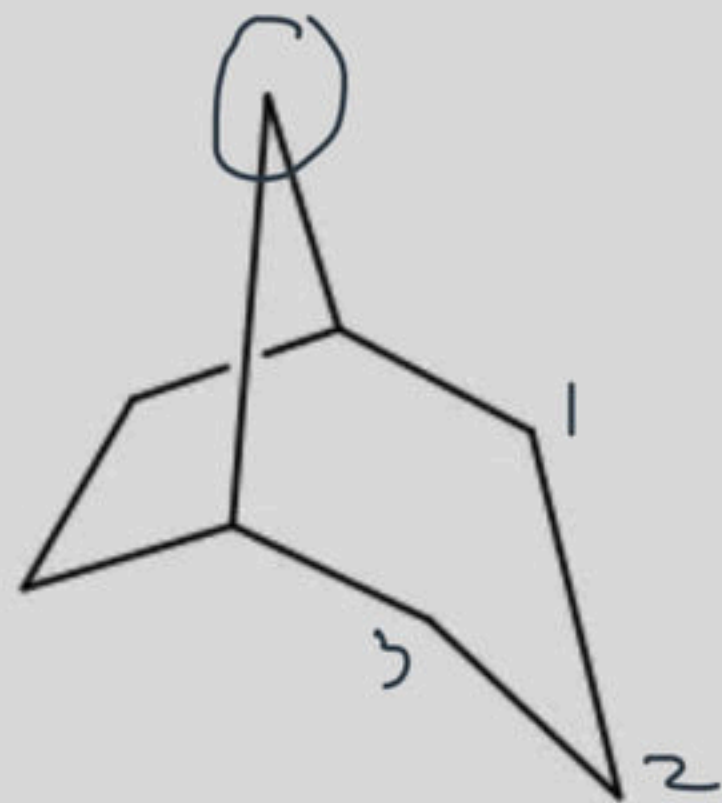
# Bridge head atoms



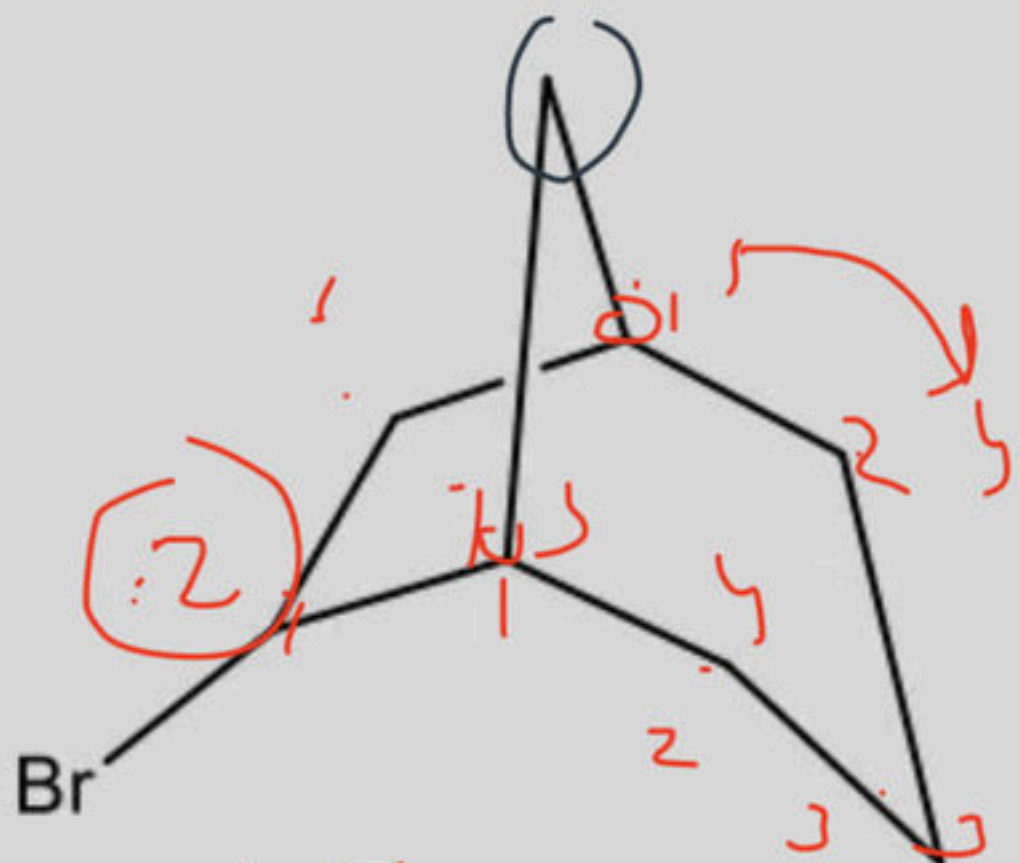
Bicyclo[3.3.1]nonane



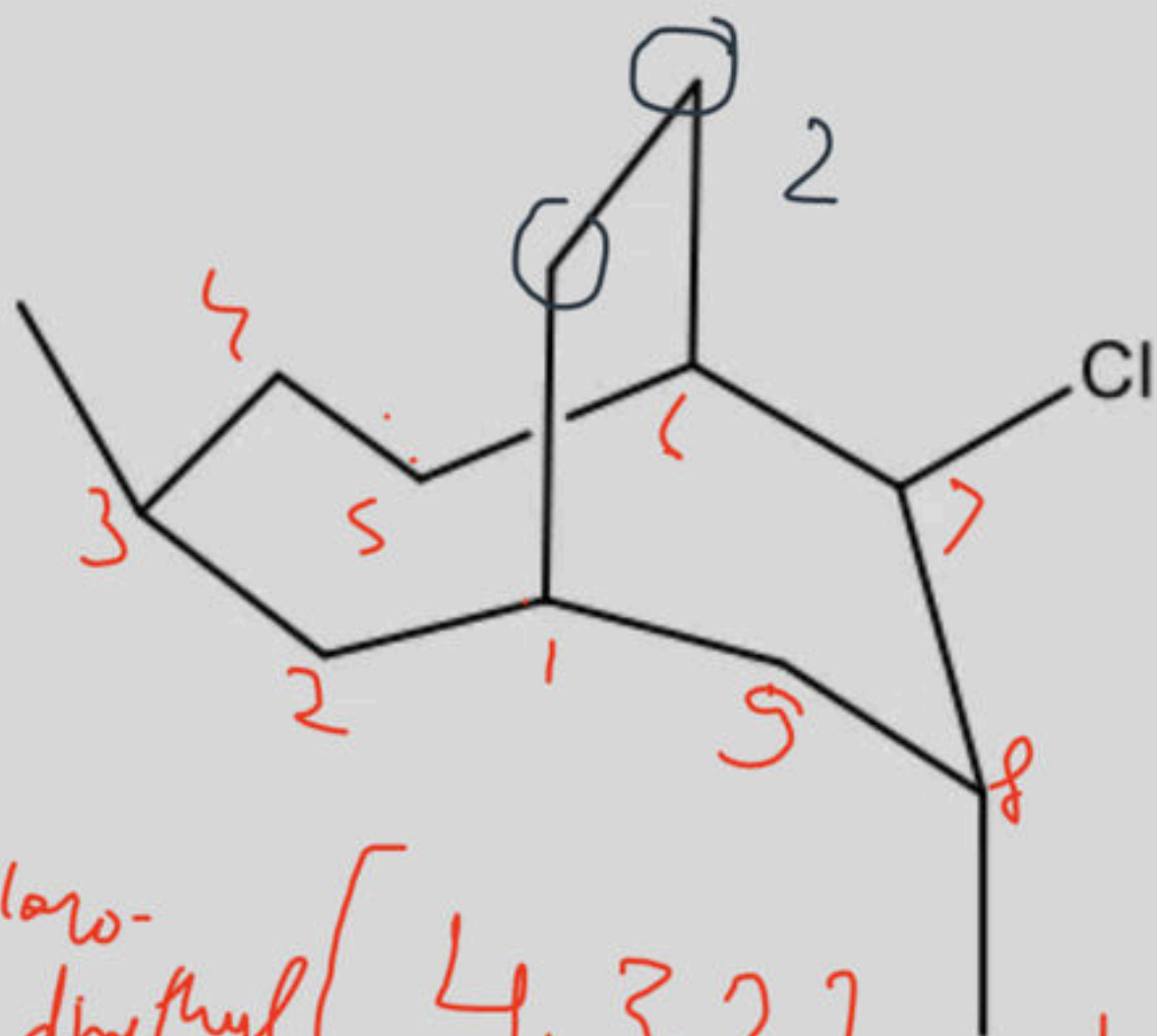
Bicyclo[3.2.1]octane



Bicyclo[3.2.1]octane



6-bromo bicyclo[3.2.1]octane



1-chloro-2,8-dimethyl bicyclo[4.3.2]undecane

~~(3+7+8)~~  
(4+8+9)



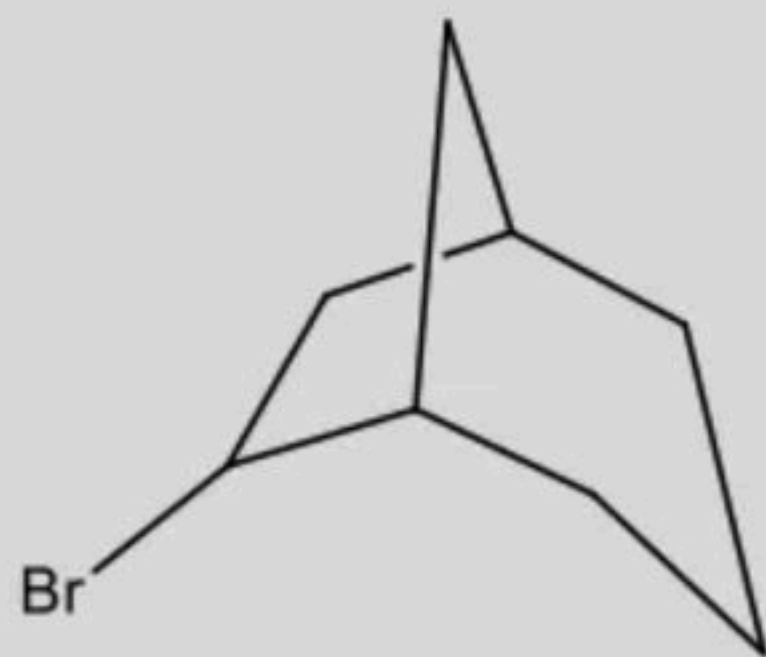
bicyclo[3.3.1]nonane



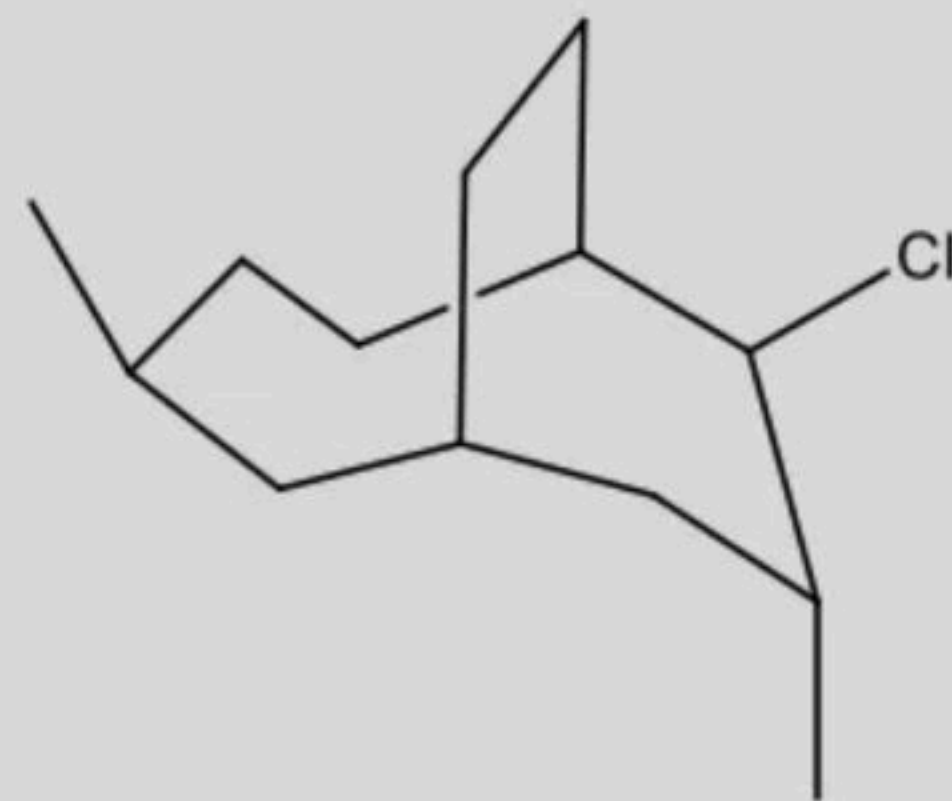
bicyclo[3.2.1]octane



bicyclo[3.2.1]octane



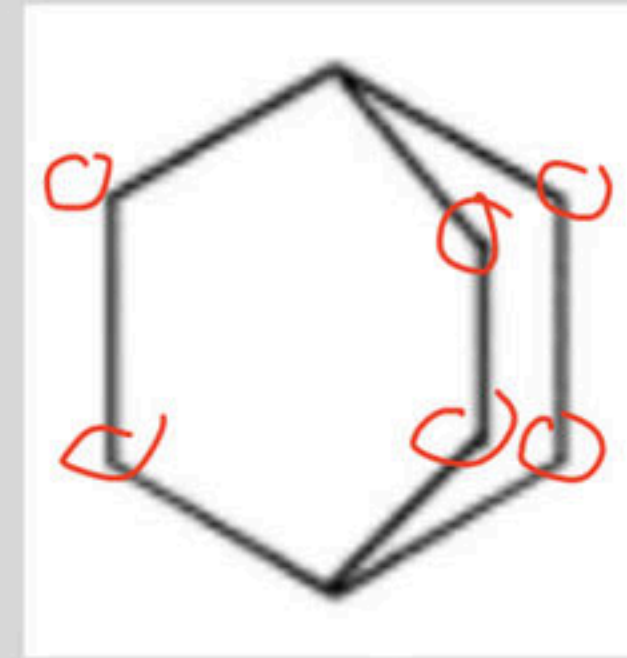
6-Bromobicyclo[3.2.1]octane

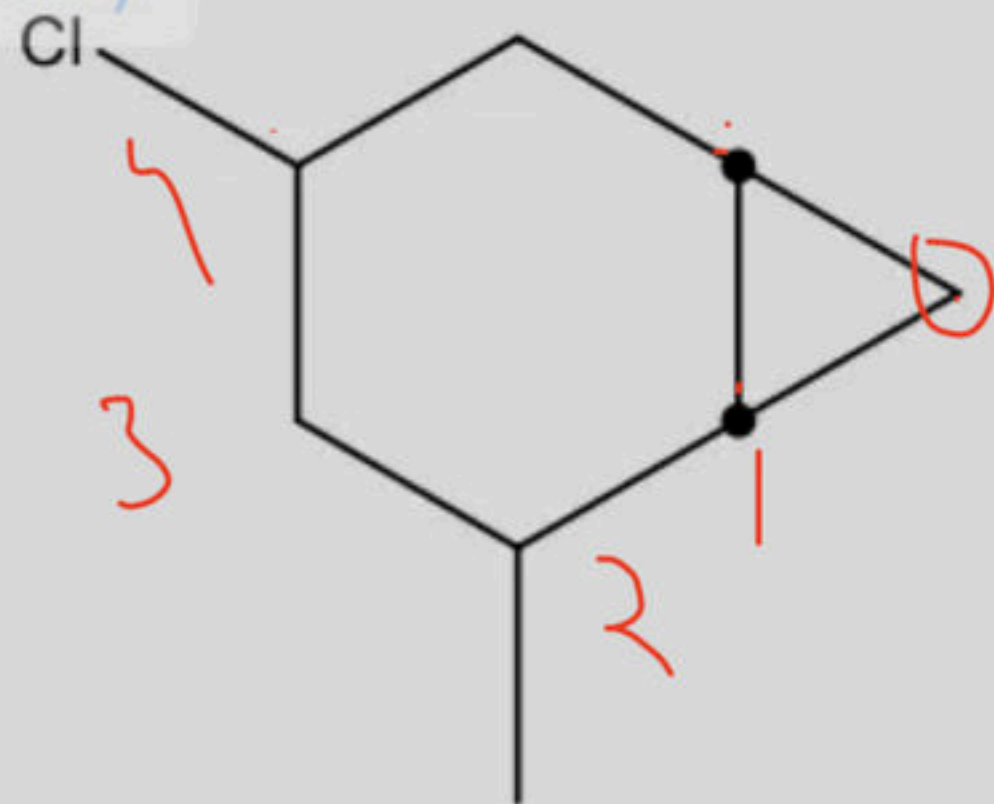


7-chloro-3,8-dimethylbicyclo[4.3.2]undecane

What is the IUPAC name of the following compound?

- a) Bicyclo(2.2.2) octane ✓
- b) Tricyclo(2.2.2) octane
- c) Bicyclo(2.2.0) octane ✗
- d) Bicyclo (2.2.1) heptane ✗





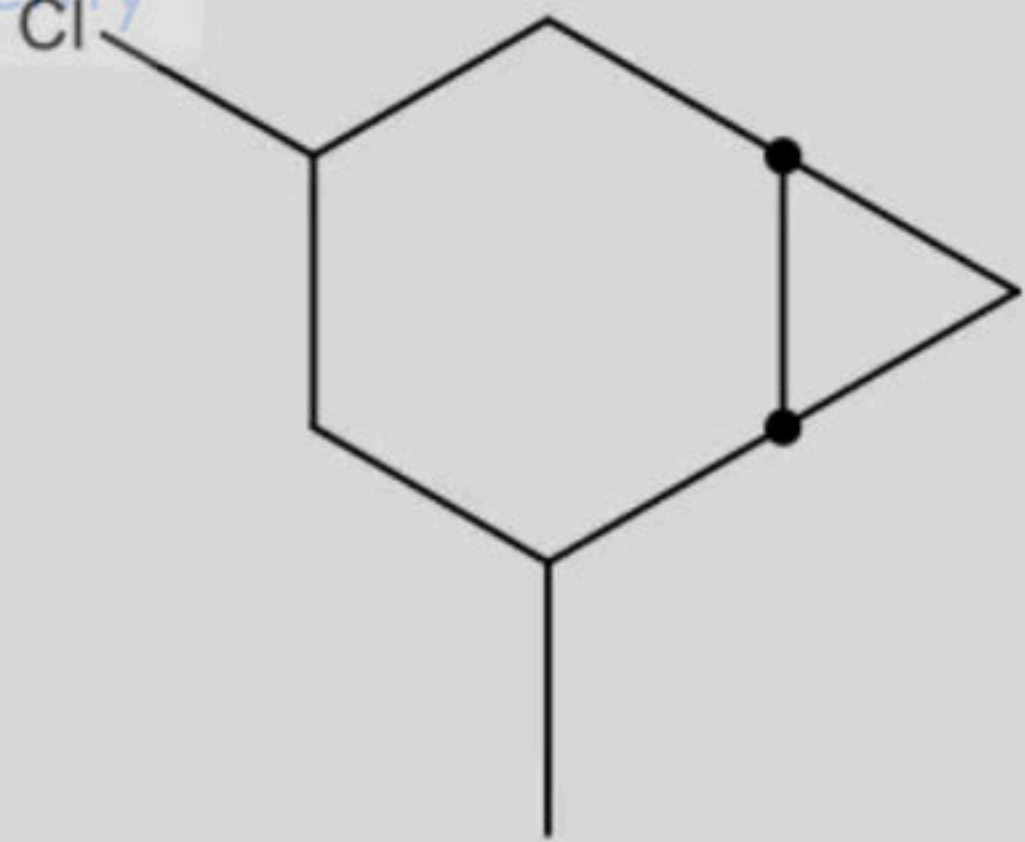
✓ A. 4-chloro-2-methylbicyclo[4.1.0]heptane ✓

B. 3-chloro-5-methylbicyclo[4.1.0]heptane

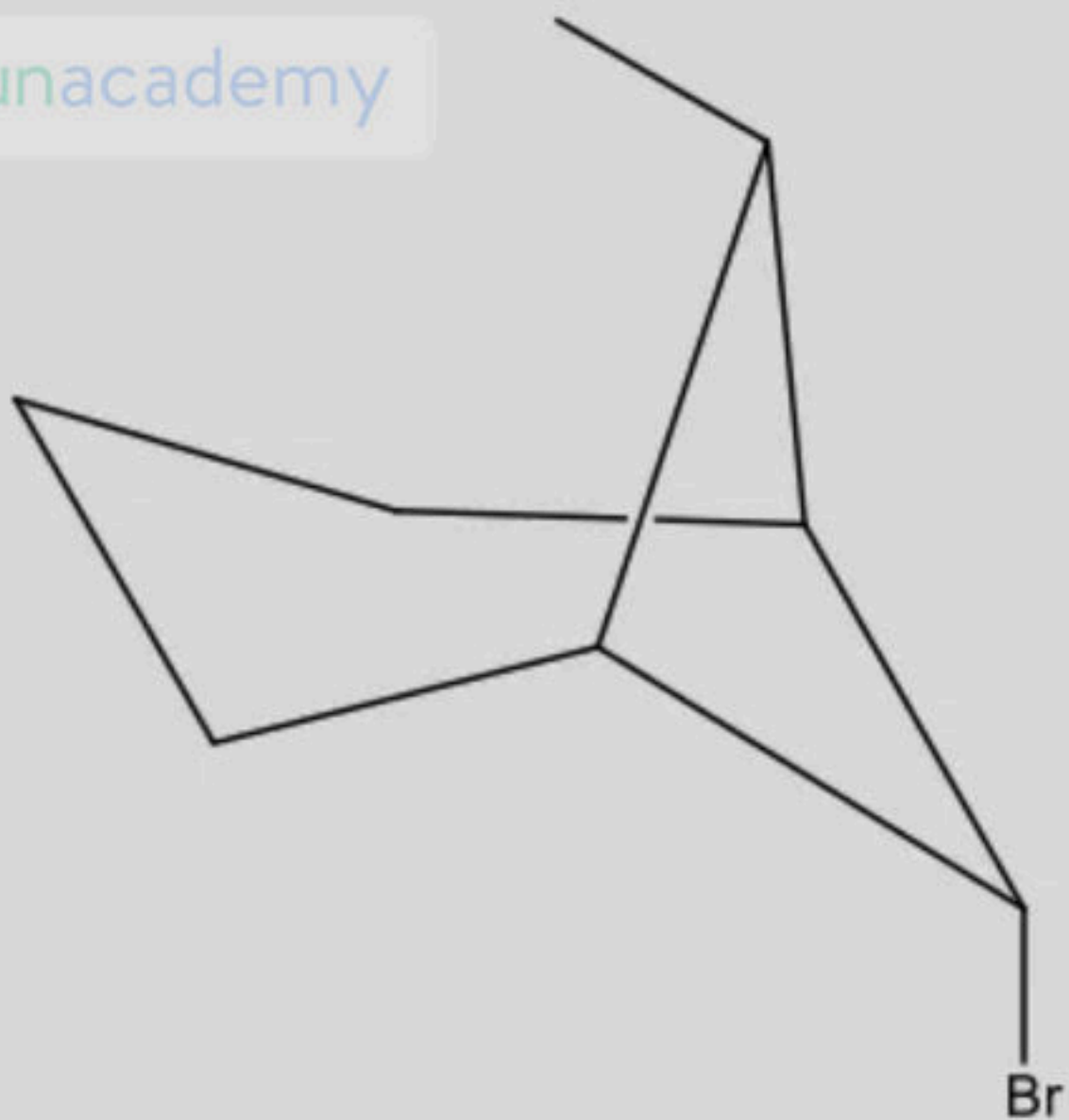
C. 6-chloro-4-methylbicyclo[4.1.0]heptane

D. 2-methyl-4-chlorobicyclo[4.1.0]heptane

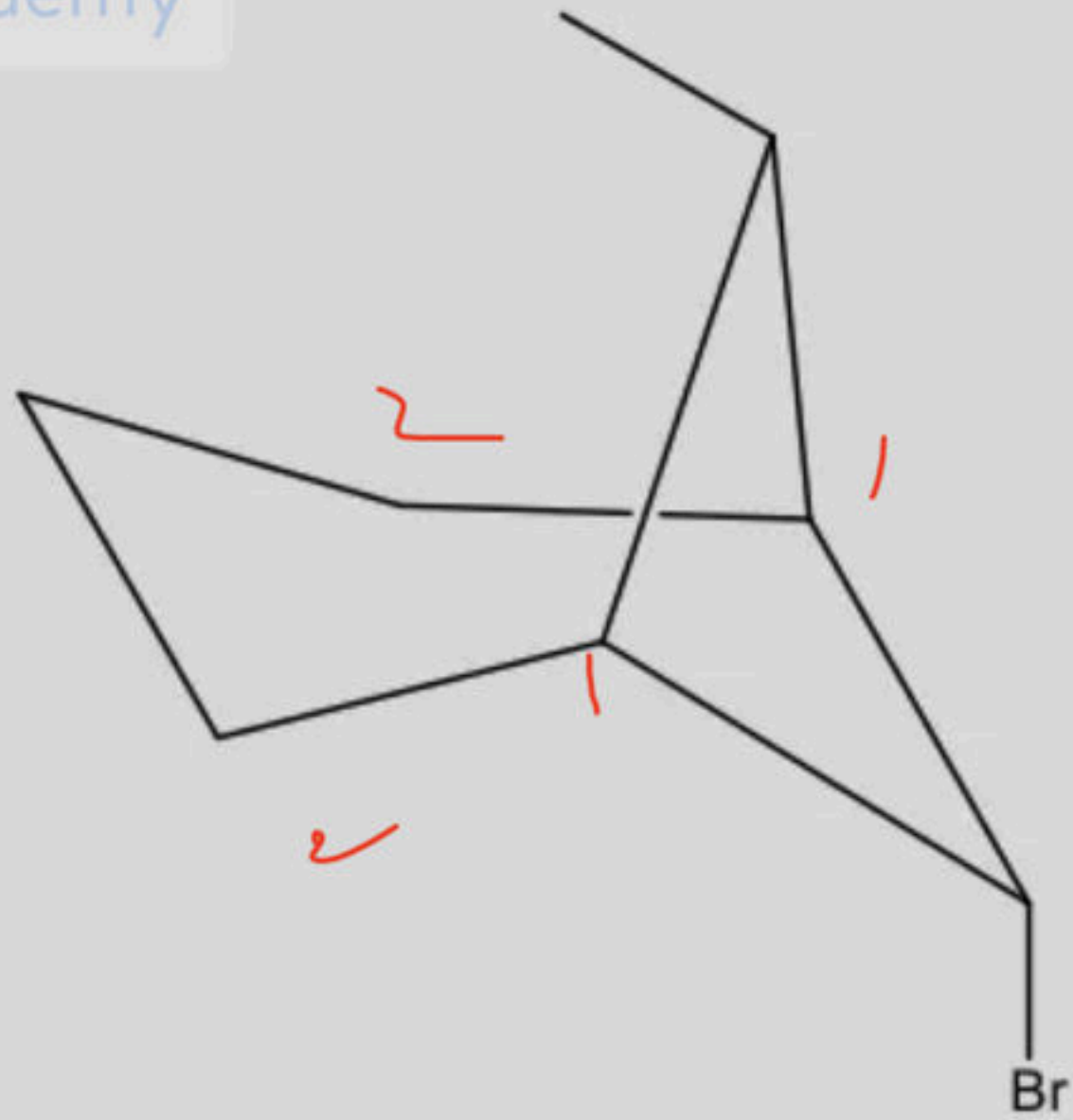
4.1.0  
 3 + 5 = 8  
2 + 4 = 6



- A. 4-chloro-2-methylbicyclo[4.1.0]heptane**
- B. 3-chloro-5-methylbicyclo[4.1.0]heptane
- C. 6-chloro-4-methylbicyclo[4.1.0]heptane
- D. 2-methyl-4-chlorobicyclo[4.1.0]heptane



- A. 7-bromo-6-methylbicyclo[3.1.1]heptane
- B. 6-bromo-7-methylbicyclo[3.1.1]heptane
- C. 2-bromo-3-methylbicyclo[3.1.1]heptane
- D. 3-bromo-2-methylbicyclo[3.1.1]heptane



A. 7-bromo-6-methylbicyclo[3.1.1]heptane

**B. 6-bromo-7-methylbicyclo[3.1.1]heptane**

~~C. 2-bromo-3-methylbicyclo[3.1.1]heptane~~

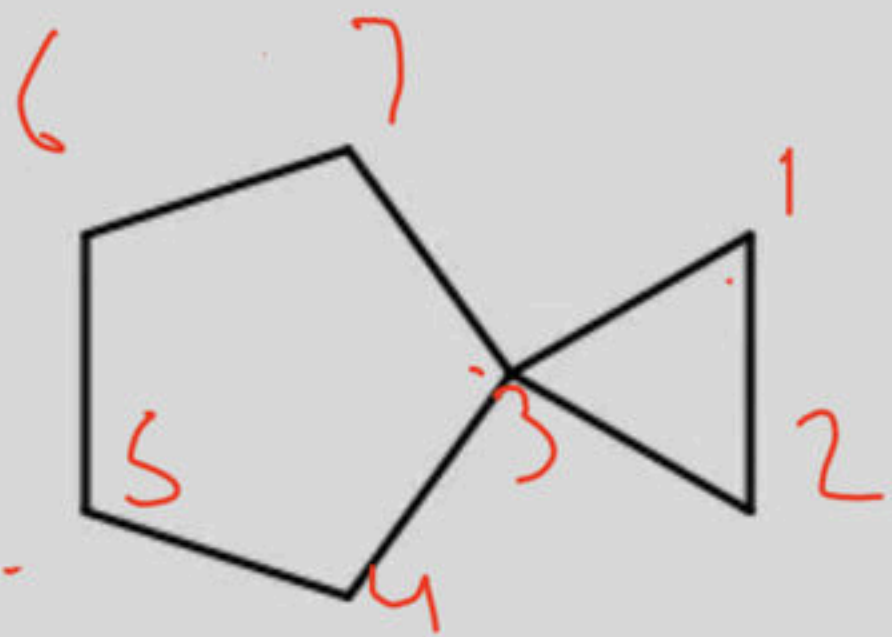
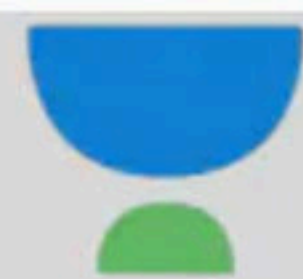
~~D. 3-bromo-2-methylbicyclo[3.1.1]heptane~~



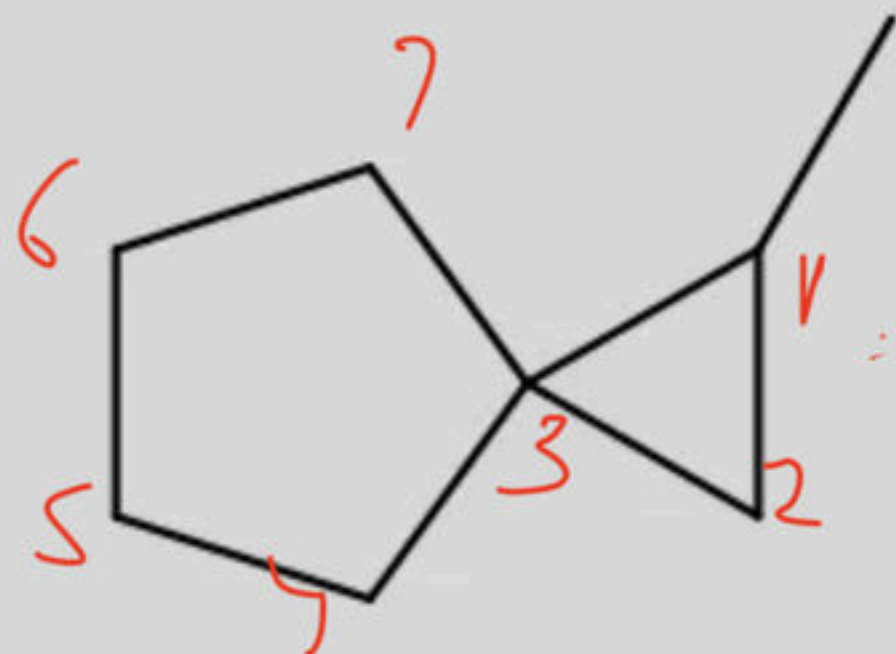
## Spirocyclic Compounds



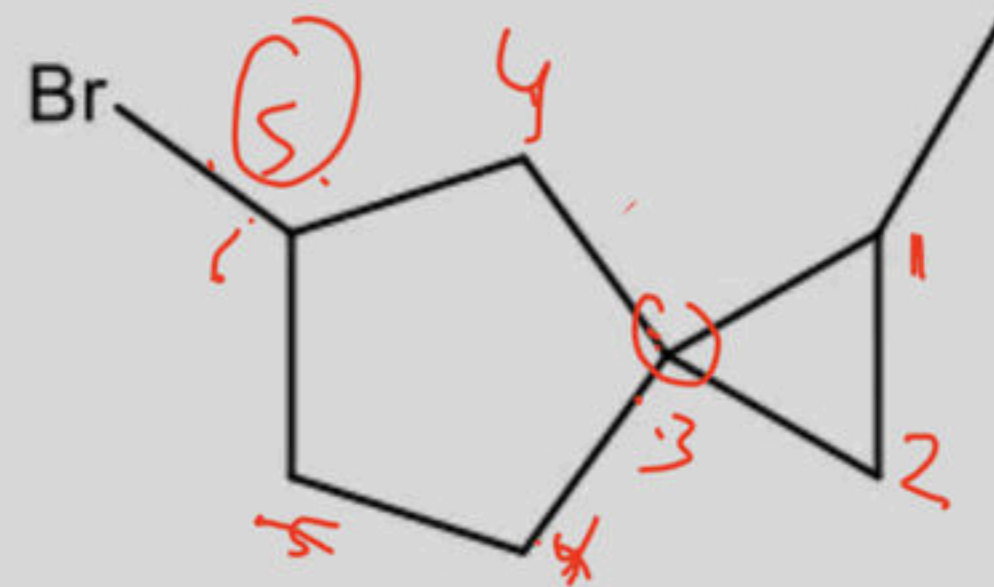
(Reverse the rules)



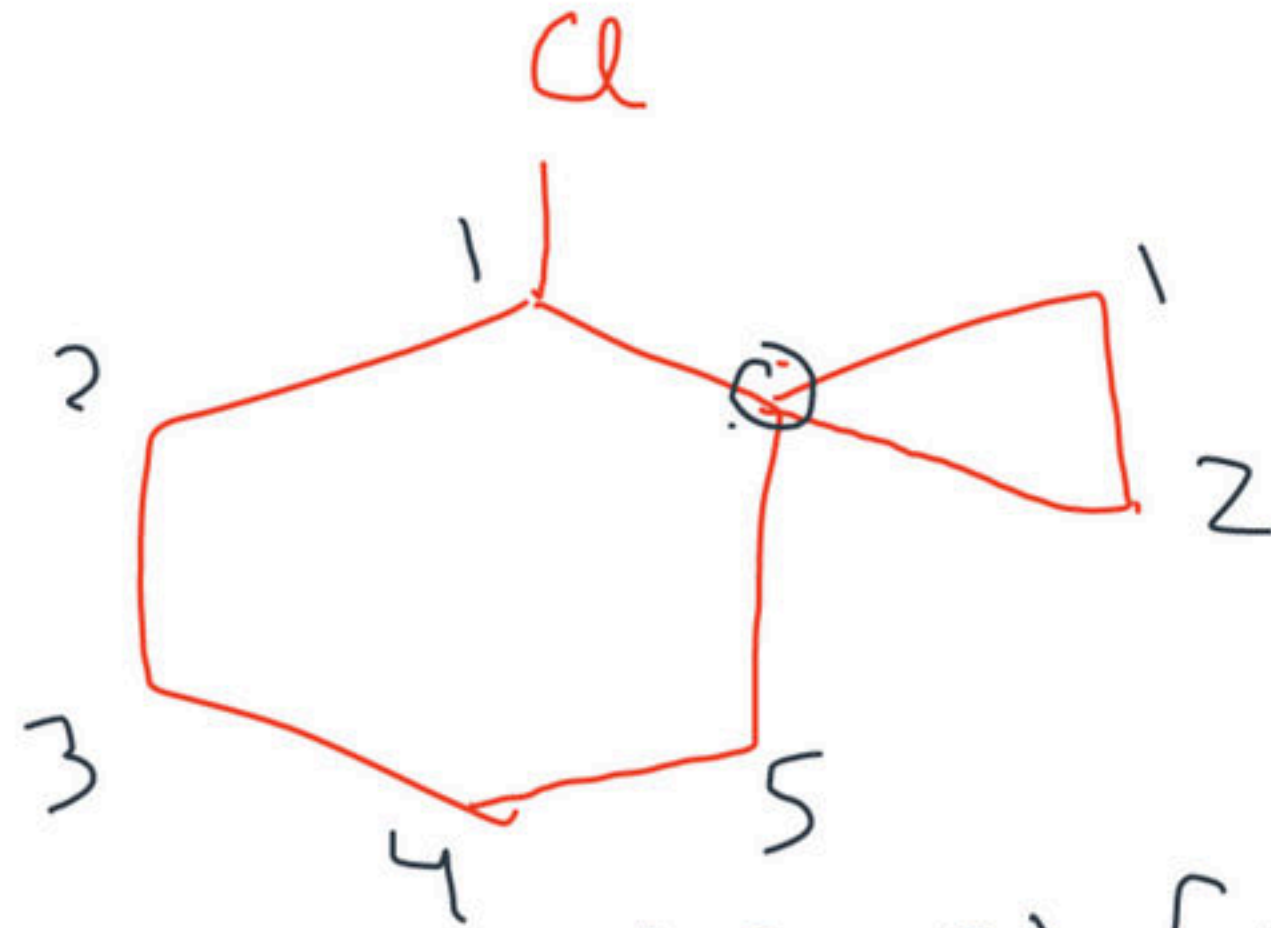
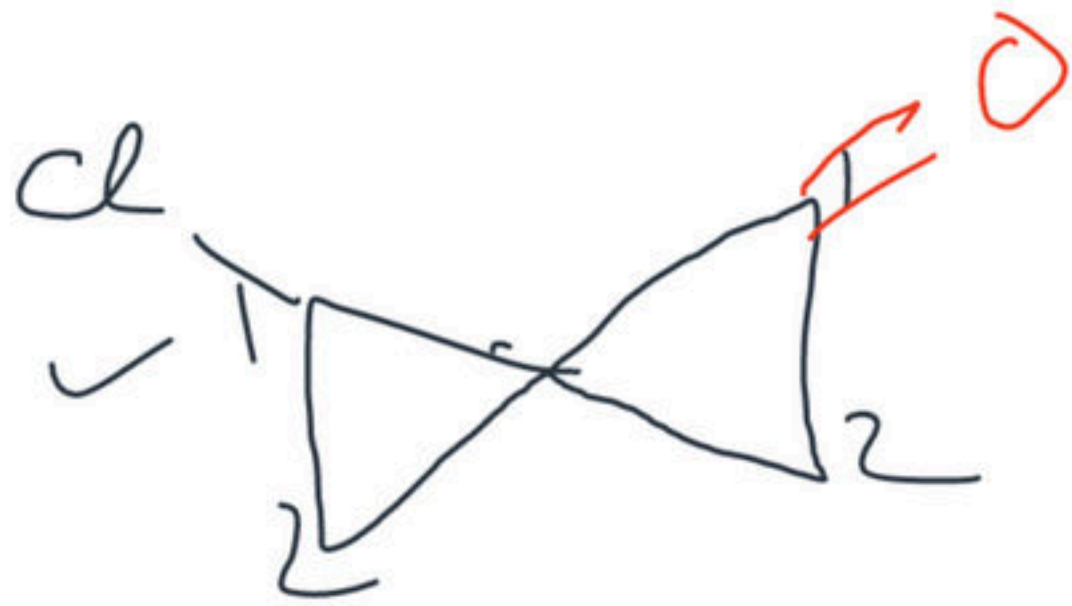
spiro[2.4]heptane



1-methylspiro[2.4]heptane

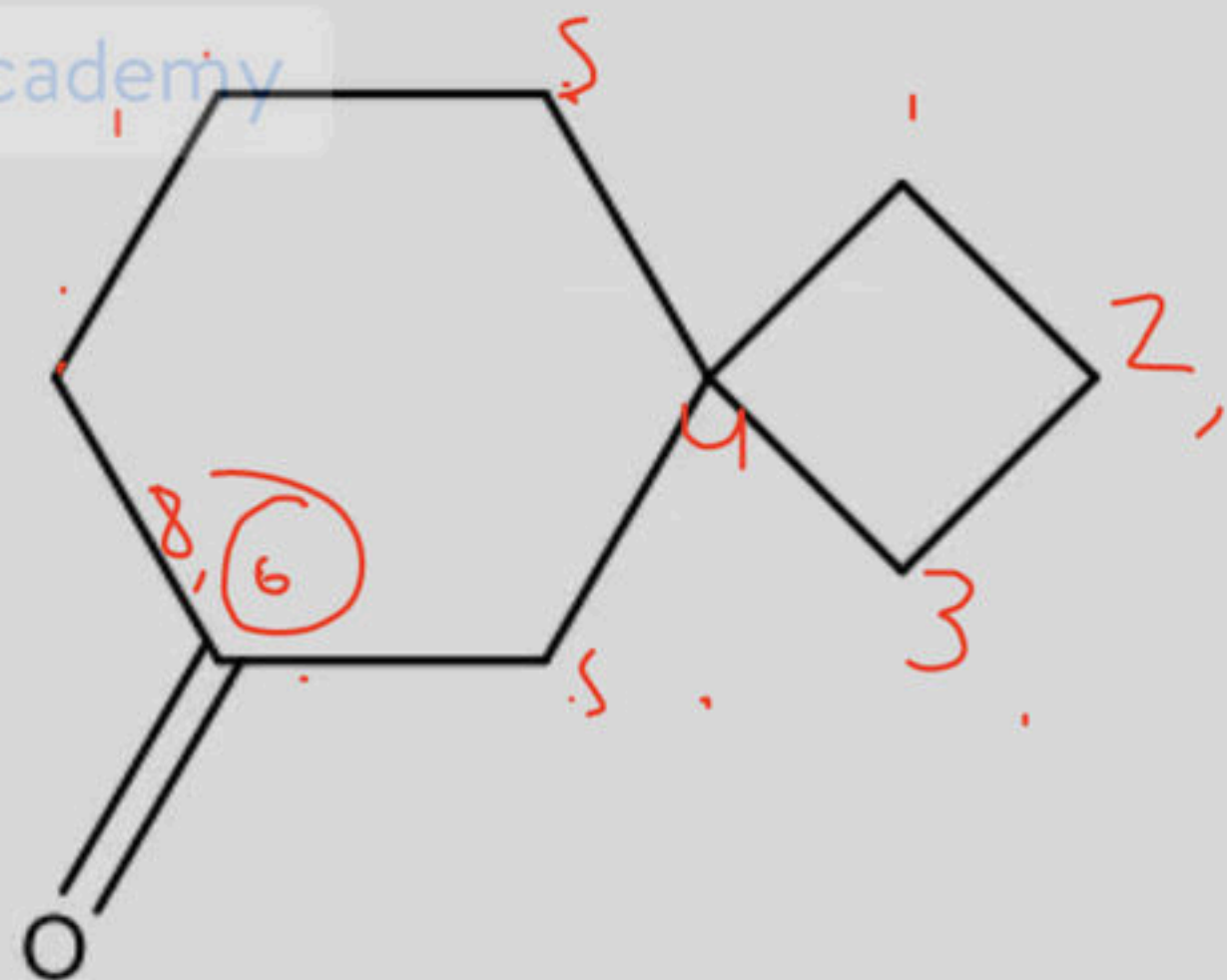


5-bromo-1-methylspiro[2.4]heptane



4-Cl spiro [2.5] octane

4-Chloro spiro [2.2] pentan-1-one

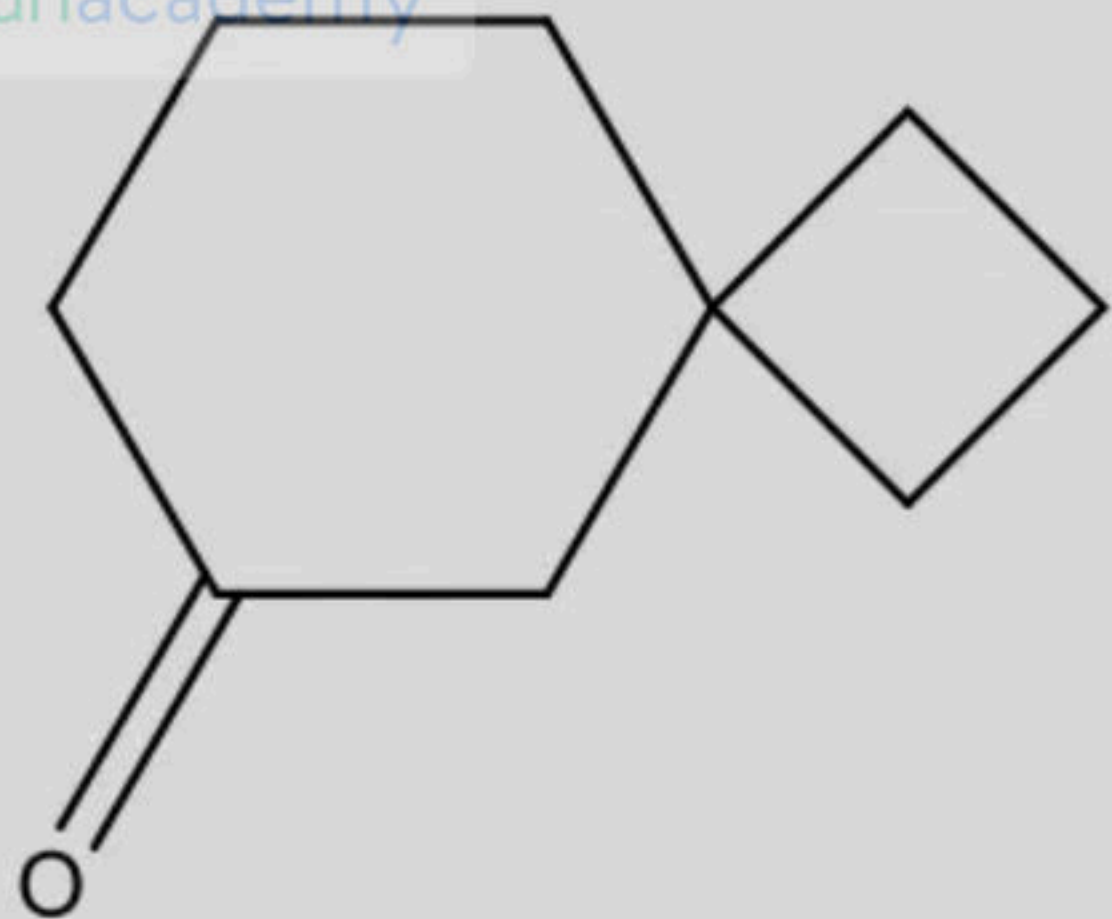


A. spiro[3.5]nonan-6-one

B. spiro[3.5]nonan-3-one ✗

C. spiro[3.5]nonan-4-one ✗

D. spiro[3.5]nonan-7-one ✗



**A. spiro[3.5]nonan-6-one** ✓

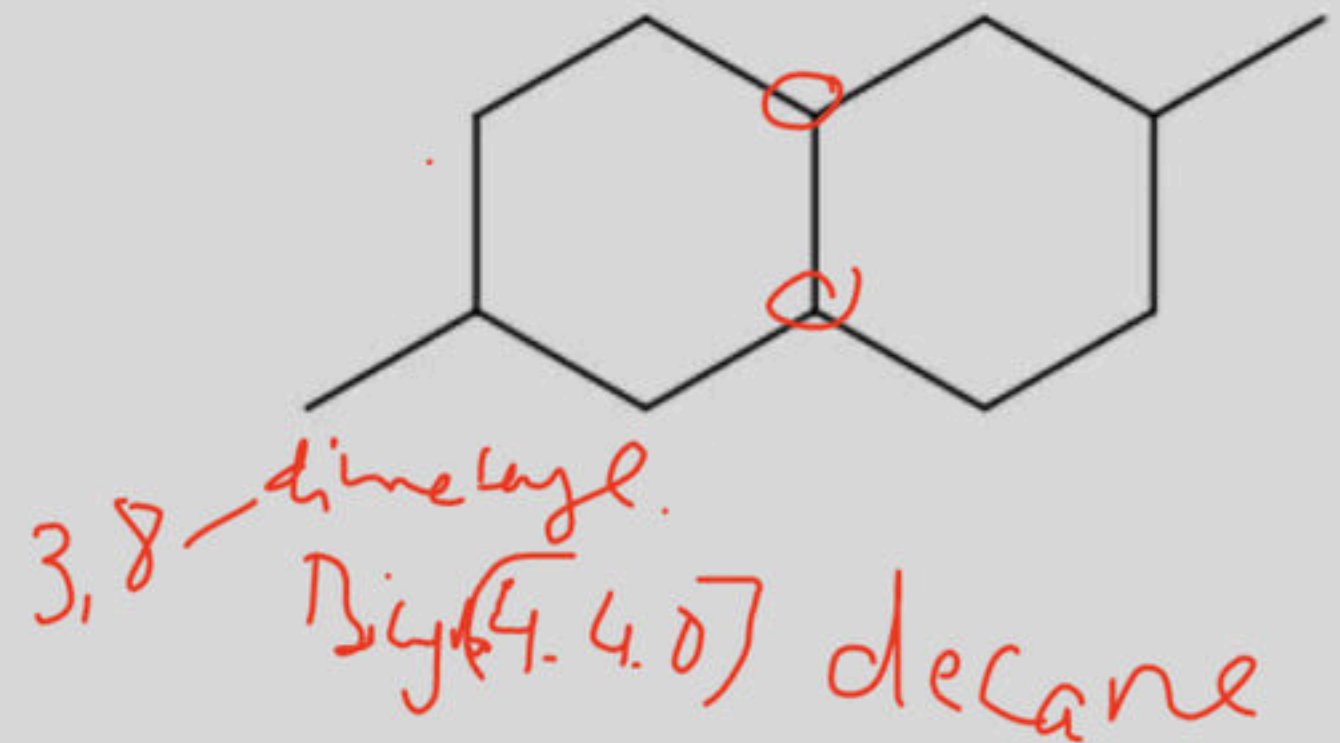
B. spiro[3.5]nonan-3-one

C. spiro[3.5]nonan-4-one

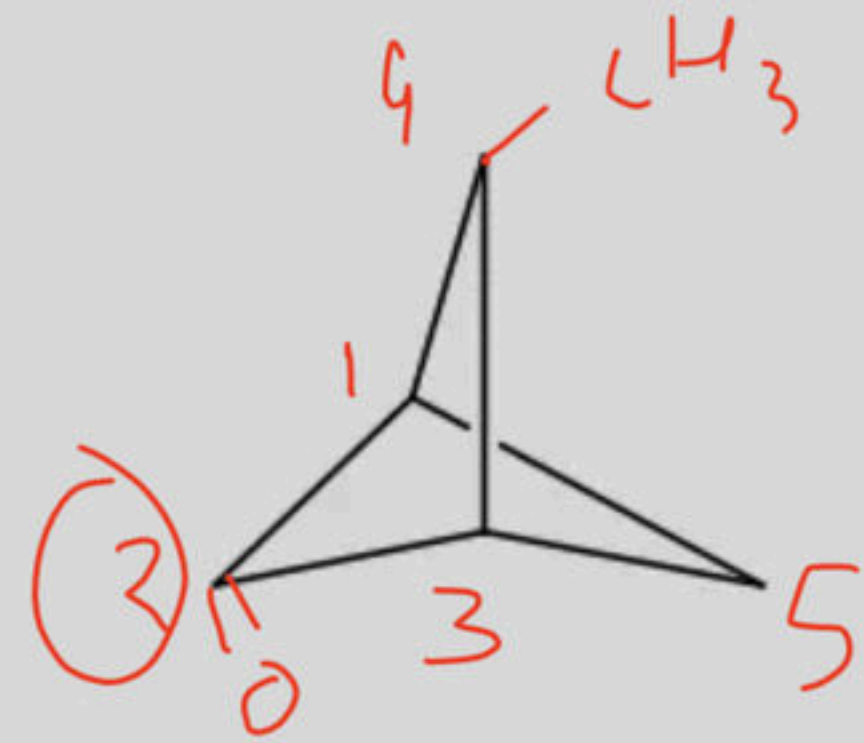
D. spiro[3.5]nonan-7-one



2,3-dimethyl spiro [4.5] decane



3,8-dimethyl. Bicyclo[4.4.0] decane



4-methyl Bicyclo [1.1.1] pentan-2-one