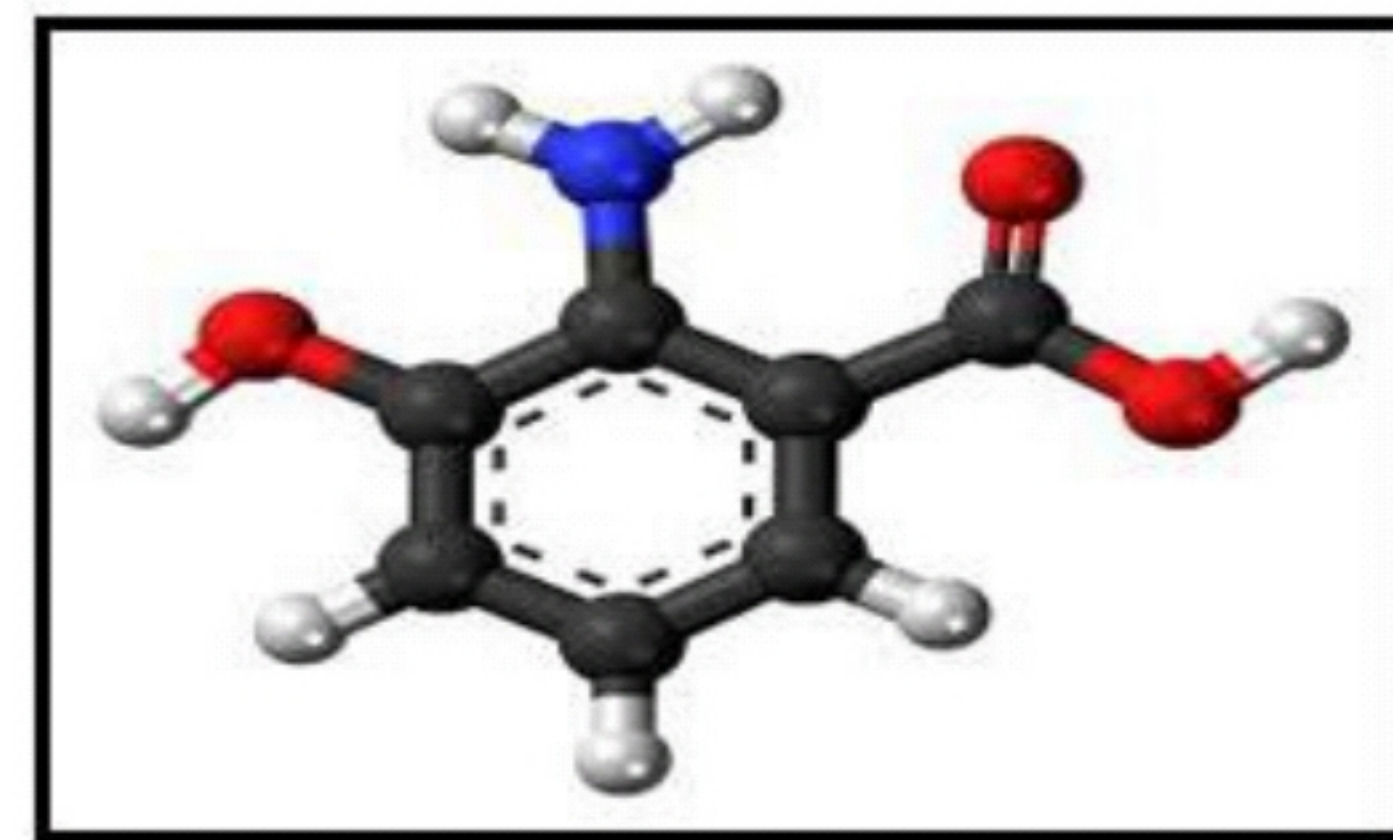
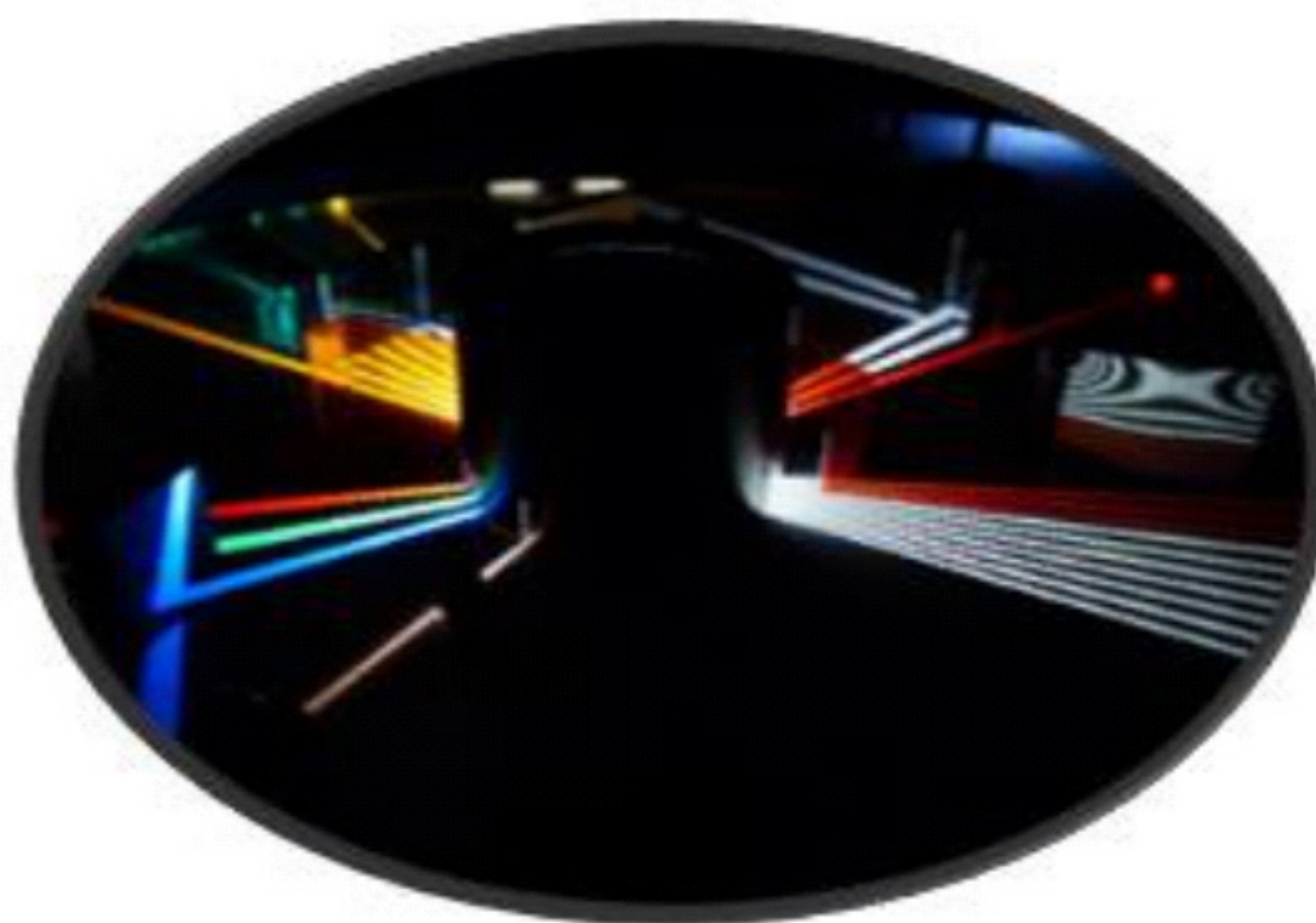
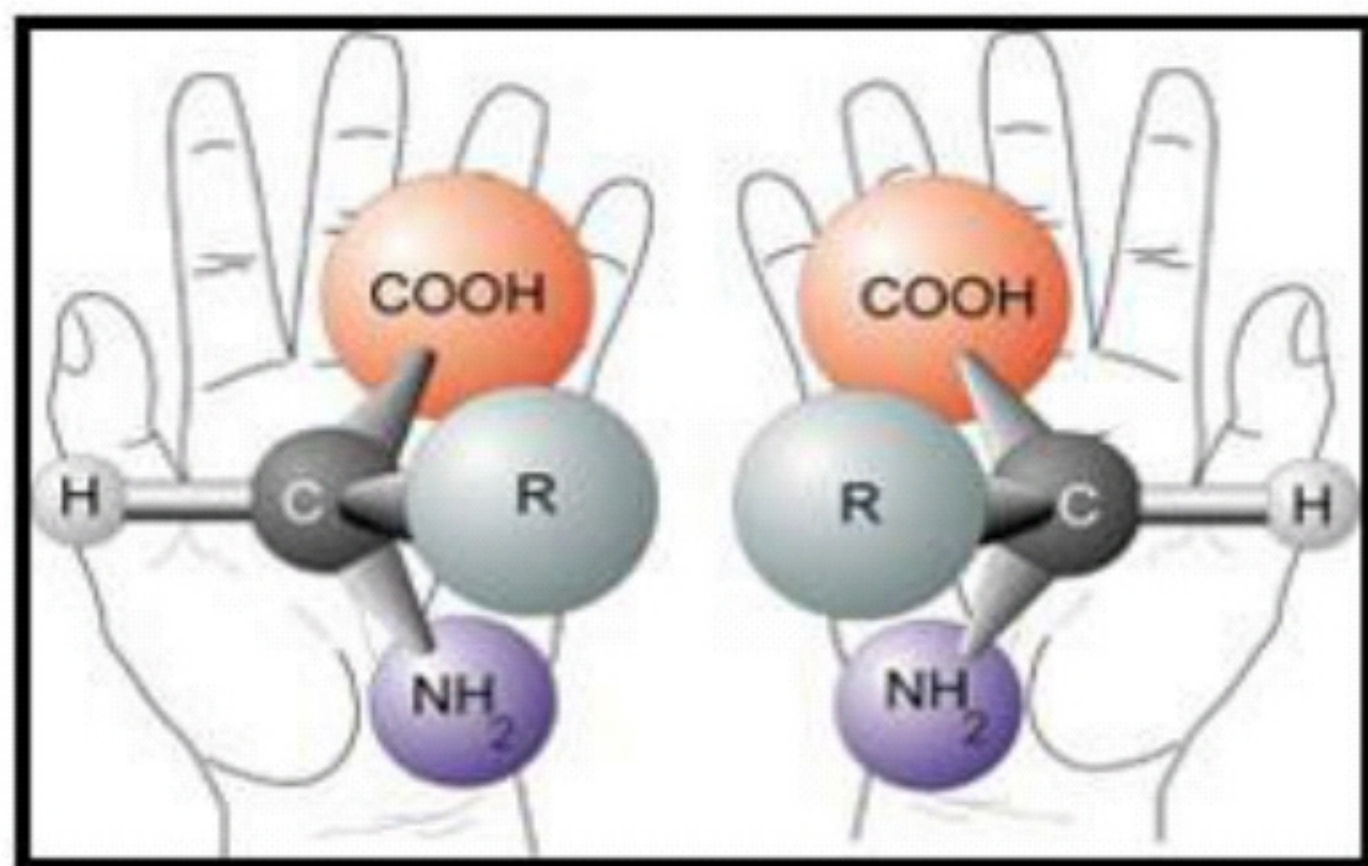
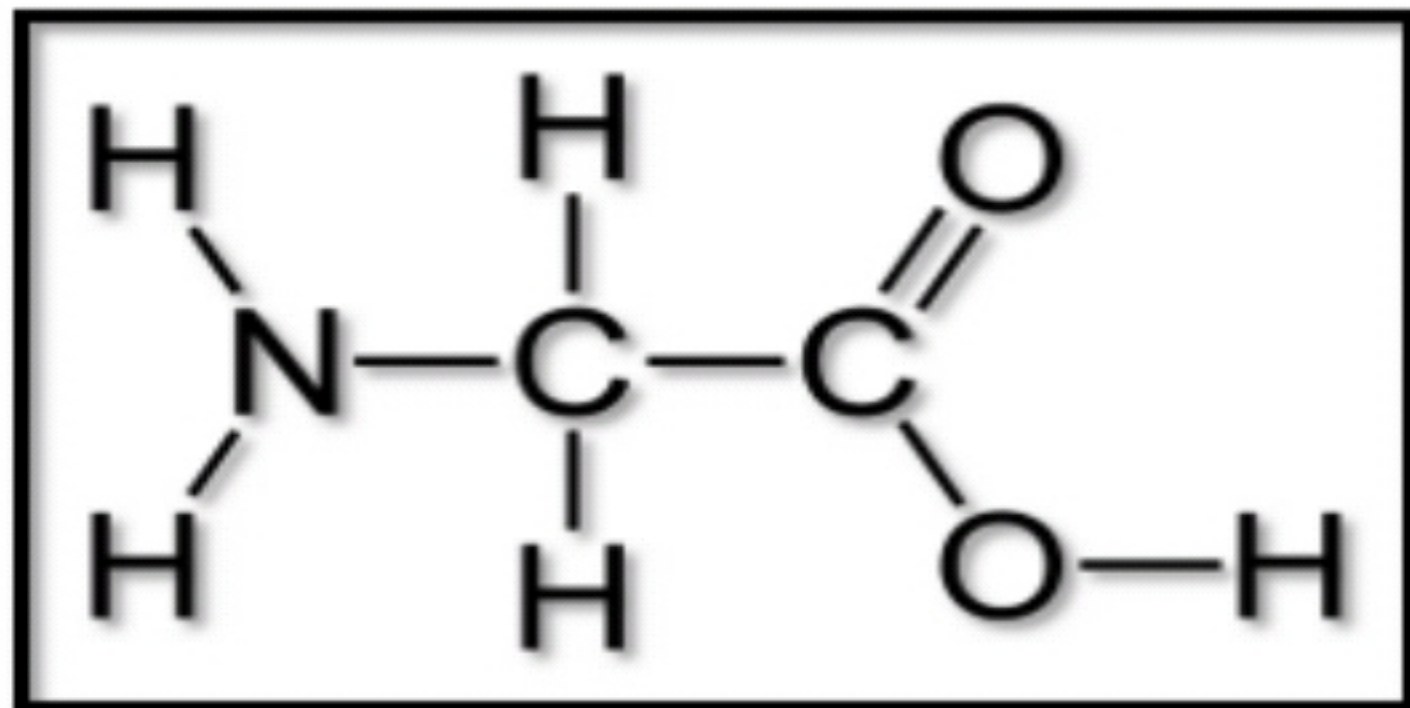


# OPTICAL PROPERTIES OF AMINO ACIDS (PART-2)

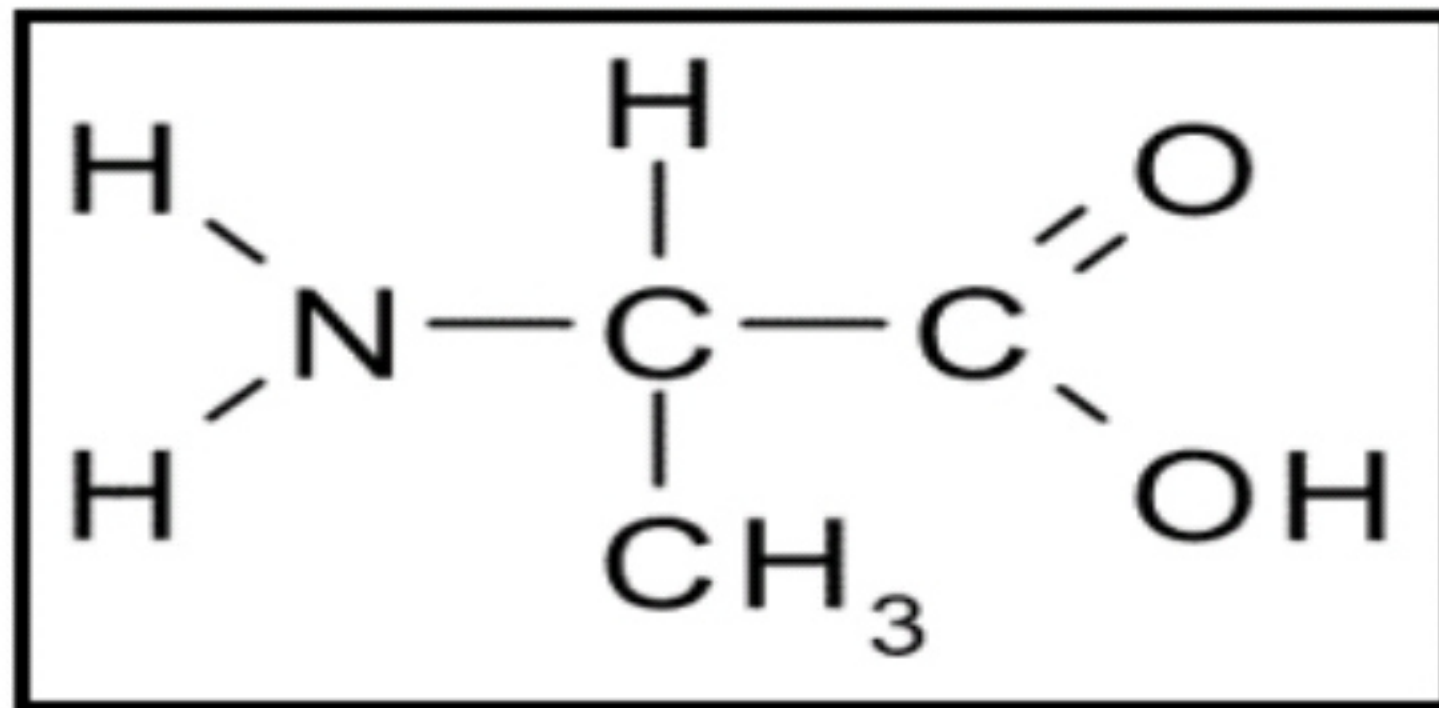




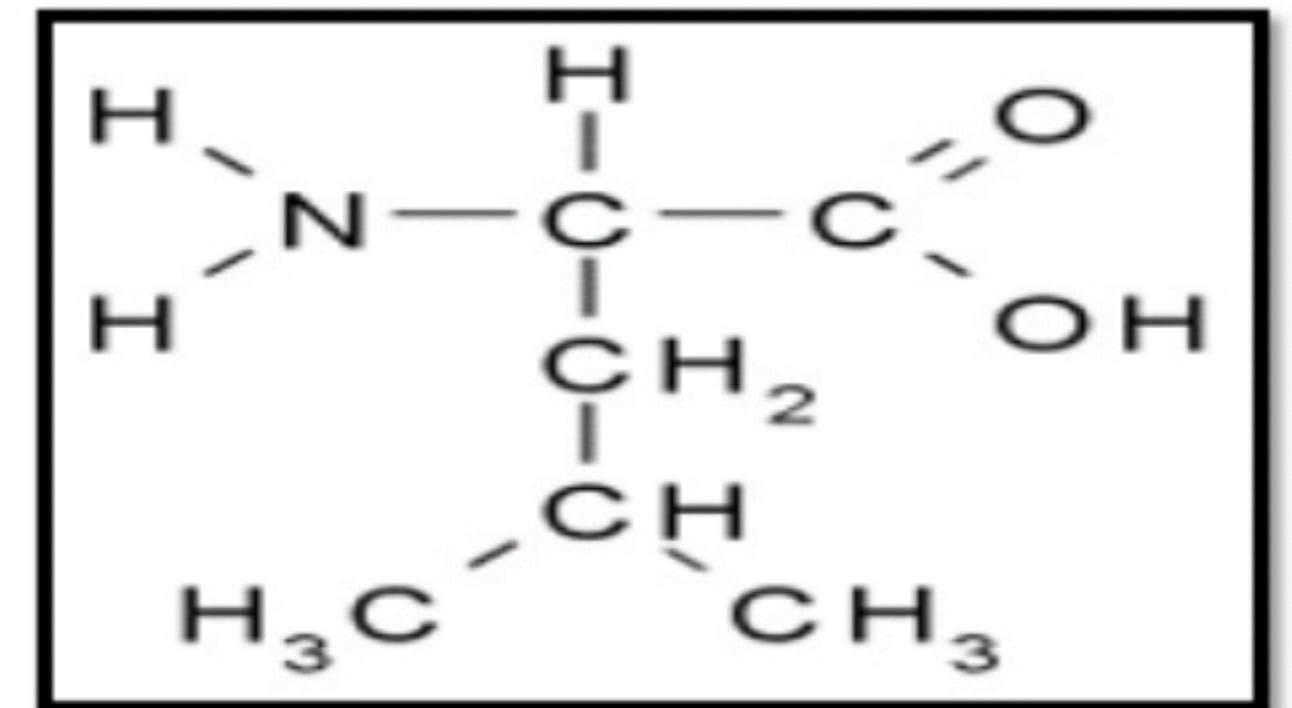
- All amino acids except **glycine** are optically active.
- **Optical active** molecules contains **chiral carbon**.
- All amino acids except glycine have chiral carbon & hence they are optically active but glycine is not.



glycine



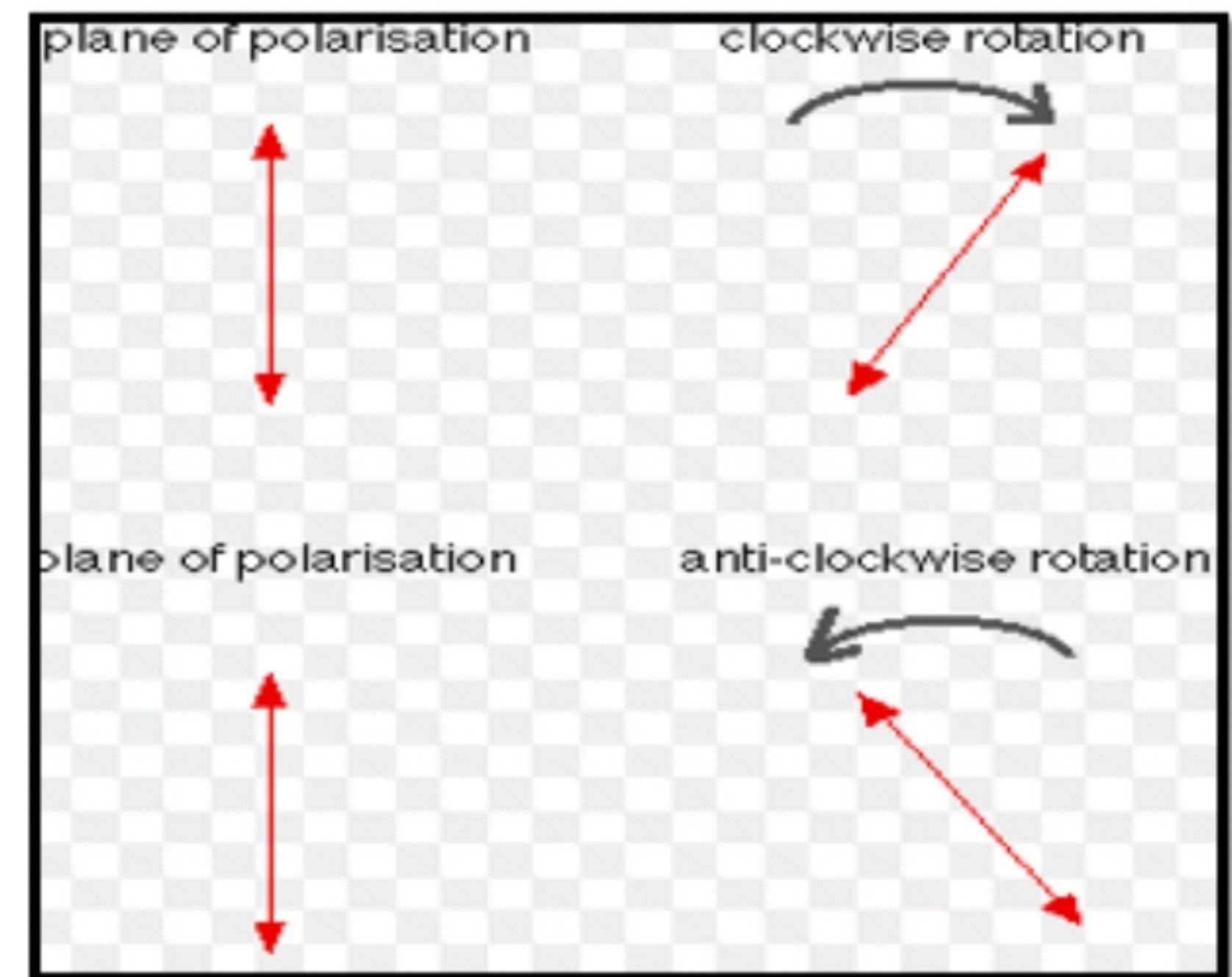
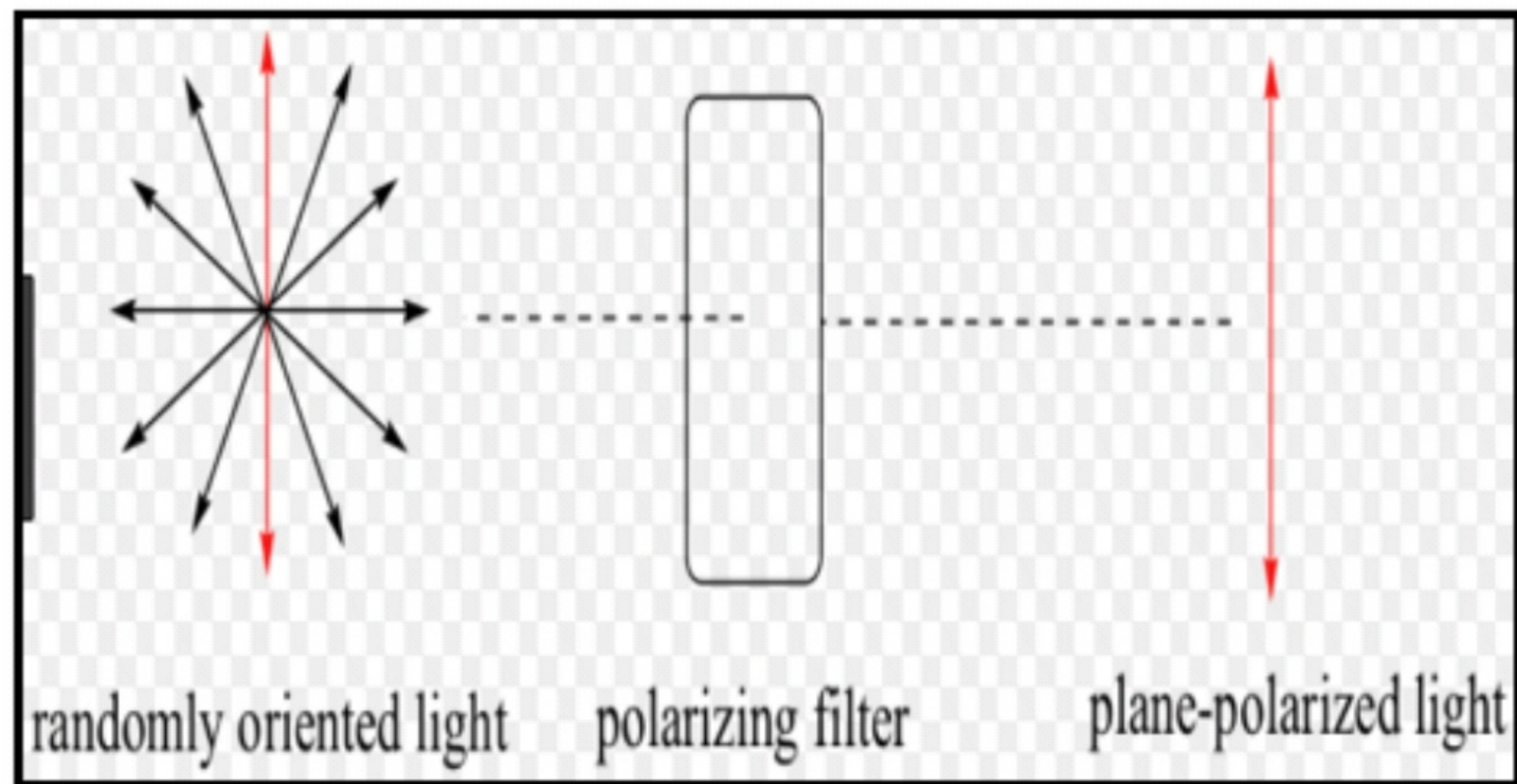
alanine



leucine



- An optically active compound can rotate the plane polarized light clockwise or anticlockwise.
- Remember what is **dextrorotatory** & **levorotatory**??





# OPTICAL ROTATION

- It is the rotation of the plane of the polarization of plane-polarized light by an optically active substance.
- It is also known as optical activity.
- Optical activity is measured by polarimeter.
- But, how can we measure it??

$$[\alpha]_{\lambda}^T = \frac{\alpha \text{ degrees}}{l \text{ dm} \times C \text{ gmL}^{-1}}$$

$\alpha$  = optical rotation

$T$  = temperature

$\lambda$  = wavelength of polarized light

$\alpha$  degrees = observed rotation in degrees

$l$  = light path length in decimeter

$C$  = concentration of solution in gm/mL



➤ A solution of L-Leucine (3.0g/50 ml of 6 N HCl) had an observed rotation of +1.81 degree in a 20cm polarimeter tube. Calculate the specific rotation of L-Leucine in 6N HCl.

$$[\alpha]_D^T = \frac{\alpha}{l \times C}$$

$$[\alpha]_D^T = \frac{+1.81}{2 \times 0.06} \quad \text{where, } l = 20 \text{ cm} = 2 \text{ dm} \text{ and } C = \frac{3 \text{ g}}{50 \text{ ml}} = 0.06 \text{ g/ml}$$

$$[\alpha] = +15.1^\circ$$



**THANK YOU..STAY POSITIVE & ENERGETIC**  
**HAPPY LEARNING!** 